



## Terminal alkyne-functionalized triazine by Sonogashira coupling: synthesis of a potential cell signalling inhibitor via click chemistry

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### ABSTRACT

Introduction of the acetylene group on the 1,3,5-triazine scaffold was studied under various conditions. We describe here a new method to functionalize a triazine ring using Sonogashira coupling between 2-benzylsulfanyl-4,6-dichloro-1,3,5-triazine and trimethylsilylacetylene to give the corresponding 2-benzylsulfanyl-4-chloro-6-(trimethylsilyl)ethynyl-1,3,5-triazine. The latter was then used to obtain phosphoric acid mono-{4-[4-(4-amino-6-benzylsulfanyl-1,3,5-triazin-2-yl)-1,2,3-triazol-1-yl]-phenyl} ester via click chemistry by Huisgen 1,3-dipolar cycloaddition reaction.

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1,2,3-Triazoles have gained increasing attention in medicinal chemistry<sup>1</sup> since the introduction of the so-called 'click chemistry' or Cu(I)-catalyzed 1,3-dipolar alkyne-azide coupling reaction, which was pioneered by Sharpless.<sup>2</sup> Protein tyrosine phosphatases (PTPs) and protein tyrosine kinases (PTKs) have been the focus of considerable drug discovery efforts in recent years, owing to their implication in regulating cell-signalling pathways. In this context, application of click chemistry recently led to the synthesis of libraries of PTP inhibitors,<sup>3</sup> and to the synthesis of the growth factor receptor-bound protein 2 (Grb2) SH2 domain-binding inhibitors designed to antagonize or modulate PTK signalling,<sup>4</sup> in addition to classical rational design of such inhibitors.<sup>5</sup>

Exploiting the triazine privileged structure<sup>6,7</sup> in medicinal chemistry<sup>8</sup> via click chemistry, which enables the construction of libraries with asymmetric diversity, would be of great interest. Alkynylation of 1,3,5-triazine derivatives has been reported under various conditions. Menicagli and Samaritani<sup>9</sup> described the synthesis of 2-(alk-1'-ynyl)-4,6-dimethoxy-1,3,5-triazines **1** by Sonogashira cross-coupling between alk-1-yne and 2-chloro-4,6-dimethoxy-1,3,5-triazine **2** in the presence of different palladium catalyst systems (Fig. 1). Furthermore, Grignard alkylation of 2,4,6-trichloro-1,3,5-triazine **3** (cyanuric chloride) was revisited,<sup>10</sup> allowing access to 2-(alk-1'-ynyl)-4,6-dialkylamino-1,3,5-triazines

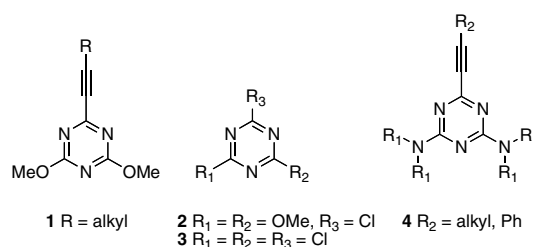


Figure 1. Alkynylation of 1,3,5-triazine derivatives.

4. Importantly, it appears that none of these papers mention the synthesis of triazines possessing the terminal acetylene group, required in our case for the Huisgen 1,3-dipolar cycloaddition reaction. To our knowledge, the introduction of this functionality, via Grignard alkylation from cyanuric chloride **3**, was only reported in a patent in 1972.<sup>11</sup>

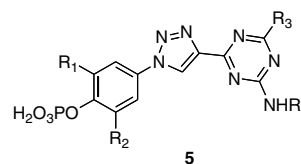


Figure 2. Triazine-functionalized, 1,4-disubstituted-1,2,3-triazoles **5**.

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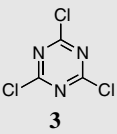
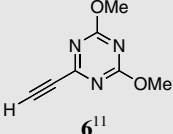
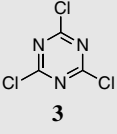
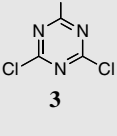
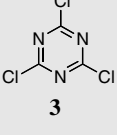
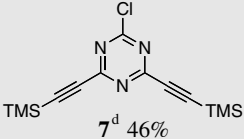
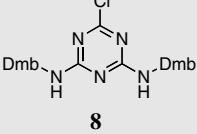
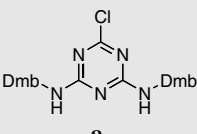
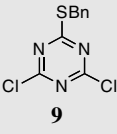
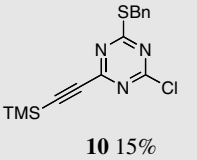
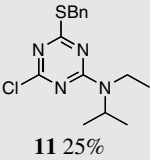
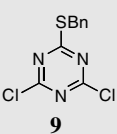
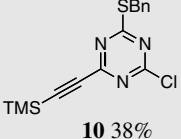
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Thus, we have designed phosphate-containing 1,4-disubstituted-1,2,3-triazoles **5** as potential cell signalling inhibitors<sup>3,12</sup> using click chemistry strategy (Fig. 2). Herein, we present the preliminary results of our investigation on the introduction of the ethynyl functionality on substituted 1,3,5-triazine derivatives, and the synthesis of phosphoric acid mono-{4-[4-(4-amino-6-ben-

zylsulfanyl-1,3,5-triazin-2-yl)-1,2,3-triazol-1-yl]-phenyl} ester **5** ( $R_1 = R_2 = \text{H}$ ,  $R_3 = \text{SBn}$ ,  $R_4 = \text{H}$ ) via click chemistry.

The introduction of the terminal acetylene functionality from **3** was demonstrated<sup>11</sup> using [(trimethylsilyl)ethynyl]magnesium bromide prepared from trimethylsilylacetylene (TMSA) and ethylmagnesium bromide (Table 1, entry 1). Addition of MeOH to the

**Table 1**  
Introduction of the trimethylsilyl ethynyl group on substituted 1,3,5-triazine derivatives<sup>a</sup>

Entry	Reaction conditions	Starting material triazine	Product yield <sup>b</sup> %
1	1. $\text{HC}\equiv\text{C-TMS}$ , EtMgBr THF, then addition of MeOH 2. Hydrolysis		 <b>6</b> <sup>11</sup>
2	$\text{HC}\equiv\text{C-MgBr}$ (1 equiv) THF, 0 °C, then addition of MeONa		Decomposition
3 <sup>c</sup>	$\text{HC}\equiv\text{C-MgBr}$ (1 equiv) $\text{Ni}(\text{acac})_2$ , $\text{PPh}_3$ , THF, 0 °C then addition of MeONa		Decomposition
4	$\text{HC}\equiv\text{C-TMS}$ (1 equiv) $\text{PdCl}_2(\text{PPh}_3)_2$ , CuI, DIPEA THF, rt, 20 h		 <b>7</b> <sup>d</sup> 46%
5	$\text{HC}\equiv\text{C-TMS}$ (1.5 equiv) Pd/C, $\text{PPh}_3$ , CuI, DIPEA MeCN, 80 °C, 20 h, sealed tube		No reaction
6	$\text{HC}\equiv\text{C-TMS}$ (1.5 or 3 equiv) Pd/C, $\text{PPh}_3$ , CuI, $\text{K}_2\text{CO}_3$ , MeCN, 80 °C, 20 h, sealed tube		No reaction
7 <sup>e</sup>	$\text{HC}\equiv\text{C-TMS}$ (1.5 equiv) Pd/C, $\text{PPh}_3$ , CuI, DIPEA MeCN, 80 °C, 24 h, sealed tube		 <b>10</b> 15%  <b>11</b> 25%
8 <sup>f</sup>	$\text{HC}\equiv\text{C-TMS}$ (2 equiv) $\text{PdCl}_2(\text{PPh}_3)_2$ , CuI, DIPEA THF, rt, 20 h		 <b>10</b> 38%

<sup>a</sup> Unless otherwise specified, reactions were carried out with Pd catalyst 0.04 equiv, CuI 0.04 equiv,  $\text{PPh}_3$  0.16 equiv, base 2.5 equiv.

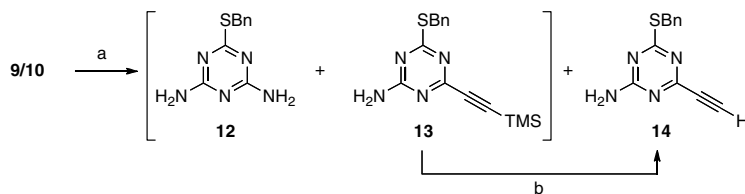
<sup>b</sup> Yields were determined by <sup>1</sup>H NMR analysis unless otherwise stated.

<sup>c</sup>  $\text{Ni}(\text{acac})_2$  0.01 equiv,  $\text{PPh}_3$  0.02 equiv.

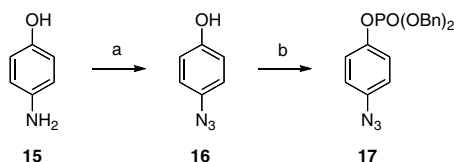
<sup>d</sup> Isolated yield.

<sup>e</sup> Flash chromatography of the crude product gave two fractions: 4-benzylsulfanyl-6-chloro-N-ethyl-N-isopropyl-1,3,5-triazin-2-amine **11** (25%) and an inseparable mixture of the starting material **9** and the desired cross-coupling product **10** (15%).

<sup>f</sup> Scaled-up reaction (8 g of **9**), **10** was inseparable from the starting triazine after flash chromatography.

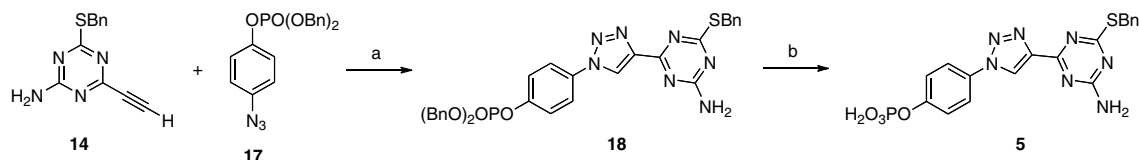


**Scheme 1.** Synthesis of alkyne **14**. Reagents and conditions: (a)  $\text{NH}_3$  (g), THF, rt, 20 h; (b)  $\text{K}_2\text{CO}_3$ , MeOH/THF (1:1), rt, 2 days, 80% (combined yield).



**Scheme 2.** Synthesis of (4-azidophenyl)dibenzylphosphate **17**. Reagents and conditions: (a) 4-aminophenol **15** diluted in HCl aq (0.5 M HCl), 0 °C, then addition of  $\text{NaNO}_2$  aq (1.5 equiv), 0 °C, then addition of  $\text{NaN}_3$  aq (1.5 equiv), 0 °C (dilution in order to obtain a 0.2 M HCl final solution), 20 h, 89%; (b)  $(\text{BnO})_2\text{P}(\text{O})\text{H}$  (2.5 equiv), DIPEA (5 equiv), DMAP (0.5 equiv),  $\text{CCl}_4/\text{CH}_3\text{CN}$  (1:5), -15 °C, 3 h, 81%.

resulting alkyne product followed by cleavage of the TMS protecting group led to 2-ethynyl-4,6-dimethoxy-1,3,5-triazine **6**. By contrast, treatment of **3** with ethynylmagnesium bromide only resulted in decomposition of the starting material (entry 2). An identical outcome was observed when Grignard alkylation was attempted with the  $\text{Ni}(\text{acac})_2\text{-PPh}_3$  system<sup>13</sup> (entry 3). Next, we examined the Sonogashira reaction<sup>14</sup> of **3** with TMSA in the presence of  $\text{PdCl}_2(\text{PPh}_3)_2$  which gave 2-chloro-4,6-bis[(trimethylsilyl)ethynyl]-1,3,5-triazine **7** in 46% yield, with only traces of the required monoethynylated product (entry 4). We then envisioned the Sonogashira coupling of 6-chloro- $N^2,N^4$ -bis(2,4-dimethoxybenzyl)-1,3,5-triazine-2,4-diamine **8**—bearing two acid-labile dimethoxybenzyl (Dmb) protecting groups for the amines<sup>15</sup>—and TMSA. Note that this triazine scaffold could lead to the synthesis of triazoles **5** featuring two nitrogen functionalizations at C4 ( $\text{NHR}_4$ ) and C6 ( $\text{R}_3$ ). Unfortunately, with  $(\text{Pd}/\text{C})/\text{PPh}_3/\text{CuI}$  as the catalytic system,<sup>9</sup> no reactions occurred (entries 5 and 6). Finally, we opted for 2-benzylsulfanyl-4,6-dichloro-1,3,5-triazine **9**<sup>16</sup> as the substrate, which allowed an orthogonal synthetic strategy, via the sulfone chemistry (i.e., oxidative activation of the thioether), for the generation of 1,3,5-trisubstituted combinatorial triazine libraries.<sup>16</sup> Under the previous reaction conditions with  $N,N'$ -diisopropylethylamine (DIPEA) as a base, triazine **9** gave rise to the formation of the desired 2-benzylsulfanyl-4-chloro-6-(trimethylsilyl)ethynyl-1,3,5-triazine **10** in 15% yield, accompanied with 25% yield of 4-benzylsulfanyl-6-chloro- $N$ -ethyl- $N$ -isopropyl-1,3,5-triazin-2-amine **11** (entry 7). The latter results from the amination of the starting material by DIPEA followed by the elimination of an isopropyl group.<sup>9b</sup> The same coupling reaction conducted at 25 °C for 24 h led to the recovery of triazine **9**. However, replacement of the catalyst  $(\text{Pd}/\text{C})/\text{PPh}_3$  by  $\text{PdCl}_2(\text{PPh}_3)_2$  afforded the desired cross-coupling product **10** in a synthetically useful 38% yield (entry 8).



**Scheme 3.** Synthesis of triazinyl-triazole phosphate **5**. Reagents and conditions: (a) ascorbic acid (0.1 equiv),  $\text{CuSO}_4$  (0.01 equiv),  $\text{H}_2\text{O}/t\text{-BuOH}$  (1:1), 30 °C, 24 h, 54%; (b) TFA 95%, rt, 2 h, 66%.

The required deprotection of trimethylsilylacetylene **10** existing as a mixture with **9** was problematic, since the reaction under TBAF conditions (TBAF, THF, rt) resulted in decomposition, and that the treatment of **10** with potassium carbonate ( $\text{K}_2\text{CO}_3$ , MeOH/THF, rt) not only yielded to the expected free acetylene group but also to concomitant substitution by methoxide anion on the 6-Cl position. We overcame this difficulty by a prior introduction of the required  $\text{NH}_2$  group from the mixture containing 2-benzylsulfanyl-4,6-dichloro-1,3,5-triazine **9** and alkyne **10** (Scheme 1): thus, reaction of ammonia with this mixture led to two fractions after flash chromatography, the less polar as an inseparable mixture of **12** and **13** (60% yield as determined by  $^1\text{H}$  NMR), and the more polar as pure isolated compound **14** (20%). Treating the mixture of **12** and **13** with  $\text{K}_2\text{CO}_3$  allowed the quantitative transformation of **13** into **14**, in a combined yield of 80%.

With compound **14** in hand, we next turned our attention to the preparation of the desired 1,4-disubstituted-1,2,3-triazole **5**.<sup>17,18</sup> First, (4-azidophenyl)dibenzylphosphate **17**<sup>19</sup> was prepared in an optimized two-step process. 4-Azidophenol **16** was obtained from 4-aminophenol by diazotization followed by treatment of the resulting diazonium salt with azide anion (89%). Phosphorylation of **16** with dibenzylchlorophosphate<sup>20</sup> (in situ prepared from dibenzylphosphite) gave **17** in 81% yield (Scheme 2).

Finally, the 1,3-dipolar Huisgen cycloaddition of alkyne **14** and aryl azide **17** was carried out to give triazole **18** in 54% yield (Scheme 3). Deprotection of **18** led to triazinyl-triazole phosphate **5** (66%) which was purified by reversed-phase C18 chromatography ( $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ ).

In conclusion, we described a study on the introduction of the ethynyl group on substituted 1,3,5-triazine derivatives via Sonogashira coupling. To the best of our knowledge, triazole **5** is the first example of click chemistry by Huisgen cycloaddition reaction applied to the privileged structure triazine.<sup>21</sup> This method will be optimized and used in the synthesis of triazine libraries. This compound and analogues will be investigated as potential cell signalling inhibitors.

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